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distance and displacement: What is  
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- Stoichiometry is the calculation of the quantities of reactants or products in a chemical reaction using the relationships found in the balanced chemical equation.

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Determine the empirical formula for a  
100.00-g sample of a compound

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having the following percent  
composition. a. 94.07% Sulfur and  
5.93% hydrogen b. 80.68%

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empirical formula for a 100.00-g  
sample of a compound having the  
following percent composition. a.

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## Chapter 12 Stoichiometry Test Answers

Chemistry: Supplemental  
Stoichiometry Problems. Directions:  
Solve each of the following problems.  
Assume excess of any reactant that  
isn't mentioned, unless otherwise .  
specified. Show your work, including  
proper units, to earn full credit. 1.  
Given the balanced equation, show  
what the following molar ratios (i.e.,  
the conversion factors) should be.

## Stoichiometry Worksheet - FREE Chemistry Materials ...

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Matter and Change • Chapter 2 1  
Data AnalysisData Analysis 1. A  
sample of aluminum is placed in a  
25-mL graduated cylinder containing

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10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

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Answer Key Answer:  $4.93 \times 10^{-5} \text{ L}$

or  $49.3 \mu\text{L}$  In Example 12.2.1 and

Example 12.2.2, the identity of the  
limiting reactant has been apparent:

$[\text{Au}(\text{CN})_2]^-$ ,  $\text{LaCl}_3$ , ethanol, and  
para-nitrophenol. Stoichiometry

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