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KS4 Rates of reaction -

reversible reactions &

dynamic equilibrium

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dynamic ...

Core Chemistry 14 - 16.  
An introduction to  
reversible reactions and

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chemical equilibria.

This page starts with a couple of simple reversible reactions, and then extends it to what happens if such a reaction is in a closed system - introducing the important idea of a dynamic equilibrium.

introduction to  
reversible reactions and  
chemical equilibria

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reactants and products.

## Equilibrium

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SECTION 18.2

REVERSIBLE  
REACTIONS AND  
EQUILIBRIUM (pages  
549 – 559) This section  
shows you how to  
predict changes in the

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equilibrium position due to changes in concentration, temperature, and pressure. It teaches you how to write the equilibrium-constant expression for a reaction and calculate its

Name Date Class

REACTION RATES  
AND EQUILIBRIUM  
18



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If a reversible reaction is exothermic/endothermic (1 mark) in one direction, it is exothermic/endothermic (1 mark) in the other.

The same (1 mark) amount of energy is transferred in each case.

(3 marks) Equilibrium

Q4: If a reversible reaction occurs in apparatus which prevents the escape of

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reactants and products,  
when would equilibrium  
be reached? A= When  
the forward and  
backward reactions (1  
mark) occur at exactly  
the same rate (1 mark).  
(2 marks) The effect of  
changing conditions on  
...

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The direction of reversible reactions can be altered by changing the reaction conditions.

## Answers

Reversible reactions and equilibria - Edexcel test

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Reversible reactions and equilibria Chemical reactions are reversible and may reach a dynamic equilibrium.

The direction of

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Reversible reactions can be altered by changing the reaction conditions....

## Answers

Reversible reactions -

Reversible reactions and equilibria ...

You need to understand how these different factors such as pressure, concentration, temperature and the presence of a catalyst

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Equilibrium  
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impact on the  
equilibrium of a  
reversible reaction. You  
need to know the  
definition of chemical  
equilibrium and  
dynamic equilibrium.  
Whether you are  
revising for the AQA  
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or you are sitting the  
exam with another  
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GCSE Chemistry

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In a reversible reaction,  
*Page 25/37*

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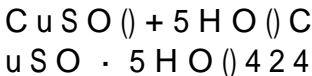
one reaction is exothermic and the other is endothermic. If the temperature is increased, the equilibrium position moves in the direction of the endothermic process....

Reversible reactions -  
BBC Bitesize  
Worksheet: Reversible  
and Irreversible

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Reactions. In this  
worksheet, we will  
practice defining  
reversibility in chemical  
reactions and identifying  
examples of reversible  
and irreversible

processes. Q1: Consider  
the following reaction:



2 s l s What is the  
enthalpy change for the  
backward reaction

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Reactions to the data  
provided in the energy  
pathway diagram?

Lesson Worksheet:

Reversible and  
Irreversible Reactions ...

Lesson for AQA

Science/Chemistry on  
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When you think of  
chemical reactions, you  
might think of them as

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irreversible, permanently changing one substance into another. While this is true in some cases, some chemical reactions are reversible, and we can take the products of the reaction and turn them back into the reactants. These reversible reactions can, under certain conditions, reach what we call

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## Equilibrium Reversible Reactions, Equilibrium, and Le Châtelier's ...

Reversible reaction:  
dehydration of blue  
copper sulfate. Blue  
copper sulfate is  
hydrated which means  
that the copper and  
sulfate ions in its crystal  
structure are  
surrounded by water

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molecules. When heated, the water is driven off, producing anhydrous copper sulfate which is a white solid. The reaction is reversible, which means that as soon as any water is around, anhydrous copper sulfate will be converted back into the blue hydrated form.



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Reversible Reactions and  
Equilibria (GCSE)

— the science hive

Reversible reactions

Chemical reactions are reversible and may reach a dynamic equilibrium. The position of equilibrium of a reversible reaction can be altered by changing the reaction conditions.

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Reversible reactions and

AQA test questions -

AQA - GCSE ...

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question Quiz

"Reversible Reactions  
and Chemical

Equilibrium" ... In a  
reaction, the side of the  
equilibrium with the  
least number of gas  
molecules in the

equation is favoured by  
pressure. (10) Changing

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Concentrations can also change the . of an equilibrium. If the concentration of reactants is increased then more . will ...

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"Reversible Reactions  
and ...

Many reactions are irreversible. But in a reversible reaction, the

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Reactions can react to produce the original reactants. At equilibrium, the concentrations of reactants and products do not change.

Reversible reactions -  
Reversible reactions -  
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About This Quiz &  
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